



Assessment grid			
Subject: Science		Year: 7	Topic/module: Particles
KS4 target direction	4	6	8(9)
Advanced	Enrichment/extension – reaching, or part of, next pathway → Features of work may include:	Enrichment/extension – reaching, or part of, next pathway → Features of work may include:	Enrichment/extension Features of work may include:
Secure <i>Students must achieve competence in all statements before being judged 'Secure'</i>	Secure The student can: <ul style="list-style-type: none">• Match particle models to the properties of a material.• Match properties of the three states of matter to the name of the state.• Recognise that different substances boil at different temperatures.• Describe how particles change in their arrangements during evaporation, condensation, and sublimation• Describe examples of diffusion• Describe simply what gas pressure• State examples of gas pressure in everyday situations	Secure The student can: <ul style="list-style-type: none">• Use the particle model to explain why different materials have different properties• Use ideas about particles to explain the properties of a substance in its three states• Explain changes of state using particle kinetics and temperature• Explain why different substances boil at different temperatures• Use a particle model to explain evaporating, condensing, and subliming• Use the particle model to explain diffusion• Describe evidence for diffusion• Use the particle model to explain gas pressure• Describe the factors that affect gas pressure	Secure The student can: <ul style="list-style-type: none">• Evaluate particle models that explain why different materials have different properties• Explain why there is a period of constant temperature during melting and freezing• Interpret melting point data.• Use the particle model and latent heat to explain boiling• Explain what occurs during sublimation and condensation using particle models• Explain the differences between evaporation and boiling• Explain the factors that affect diffusion• Explain, using particle diagrams, what happens to gas pressure as the temperature increases
Developing	Mostly secure – one or more gaps For example:	Mostly secure – one or more gaps For example:	Mostly secure – one or more gaps For example:
Beginning	Significant gaps	Significant gaps	Significant gaps